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the line at the left of

each statement, write

the letter of the choice

that best completes

the statement or

answers the question.

1) Chemistry is a

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Answers Matter And  
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With Answer for  
Chemistry chapter 1  
(Fundamental  
Concepts) In this  
topic, Student should

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be able to : a) Define relative atomic, molecular and formula masses, based on the  $^{12}\text{C}$  scale and concept of isotopes. b) Explain mole in terms of the Avogadro's constant.

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covalent bonds. atoms  
are held together by

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sharing electrons. ... a well-test explanation for a broad set of observations. ... elements in the modern periodic table are arranged in order of increasing \_\_\_\_\_ and in groups according to their \_\_\_\_\_

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chemical. mass.

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## **Chemistry Chapter 7**

### **Section 1 Review**

#### **Answers**

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Chapter Test Chapter:

Acid-Base Titration and

pH PART I In the space

provided, write the

letter of the term or

phrase that best

completes each

statement or best

answers each question.

\_\_\_\_\_ 1. The pH scale

generally ranges from

a. 0 to 1. b. 1 to 1. c. 0

to 7. d. 0 to 14. \_\_\_\_\_ 2.

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3.1 discusses some of the most fundamental assertions of chemistry, including the law of conservation of mass, the law of definite proportions, and the law of multiple proportions. It also gives a ...

### **3.1 The Atom**

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Review CHAPTER 5:  
THE PERIODIC LAW The following pages contain the bulk (but not all) of the information for the chapter 5 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc.

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Chapter 10 Section 1

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PLAY. Match. Gravity.

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Terms. fallacychan.

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